

## cis-Clerodane-Type Furanoditerpenoids from *Tinospora crispa*

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Received September 11, 2009

Phytochemical and cytotoxicity investigations on organic solvent extracts of the aerial parts of *Tinospora crispa* have led to the isolation of 15 *cis*-clerodane-type furanoditerpenoids. Of these, nine compounds (**1–9**) were found to be new. Spectroscopic assignments of a previously reported compound, borapetoside A (**13**), were revised on the basis of HMQC and HMBC correlations. No discernible activity was observed when compounds **10–13** were subjected to evaluation in cytotoxicity assays against human prostate cancer (PC-3) and the normal mouse fibroblast (3T3) cell lines.

*Tinospora crispa* (Willd.) Hook. f. & Thomson [syn.: *Tinospora crispa* (L.) Miessner, ex Hook. f. & Thomas, *Menispermum crispum* L., *Tinospora nudiflora* Kurz., *Tinospora rumphii* Boerl.] is a tree belonging to the family Menispermaceae. This species is used widely in Thailand, Malaysia, and Indonesia as a bitter tonic for the treatment of jaundice, rheumatism, urinary disorders, and intermittent fever.<sup>1,2</sup> *Tinospora* species are among the most valuable herbs in Ayurvedic medicine and are well known for their antiallergic, antidiabetic, anti-inflammatory, antioxidant, antispasmodic, hepatoprotective,<sup>3–5</sup> and immunomodulatory activities.<sup>6</sup> A number of clerodane-type furanoditerpenoids have also been reported with activity against human oral epidermoid, medulla, and colon cancer cell lines.<sup>7</sup> Several studies have been conducted on the constituents of the genus *Tinospora*, and a variety of compounds have been isolated including furanoditerpene lactones, steroids, flavonoids, lignans, alkaloids, and phenolics.<sup>8</sup> Among them, clerodane-type furanoditerpenes are commonly found in the genus *Tinospora*.<sup>5–8</sup>

In the present investigation, nine new (**1–9**) and six known (**10–13**, borapetosides F and G) clerodane-type furanoditerpene lactones have been isolated and characterized.<sup>8,10–12</sup> Moreover, the <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts for Me-19 and Me-20 in compound **13** were also reassigned. Compounds **10–13** were tested for their cytotoxic activities.

### Results and Discussion

The air-dried aerial parts of *T. crispa* were powdered and extracted with 80% aqueous methanol. The methanol extract (280 g) was partitioned sequentially into *n*-hexane, EtOAc, *n*-BuOH, and water. The ethyl acetate and *n*-butanol fractions were subjected to repeated silica gel, Diaion HP-20, Sephadex LH-20, and ODS RP-18 column chromatography, while final purification was carried out by preparative HPLC, to afford compounds **1–13** and borapetosides F and G. The known compounds **10–13** were identified as borapetosides B, C, D, and A, respectively, along with borapetosides F and G, on the basis of the comparison of their physical and spectroscopic data with those reported in the literature.<sup>8,11,12</sup>

Compounds **1–9** were identified as clerodane-type furanoditerpenes on the basis of their characteristic spectroscopic features.<sup>13</sup> The EIMS of these compounds exhibited a characteristic fragmentation at *m/z* 81 as a result of bond fission at C-11/C-12. Fragment ion peaks at *m/z* 94 and 95 were due to cleavage of C-9/C-11 bond.

The UV spectra showed three absorption bands in a range of 200–225 nm. The IR spectra (KBr) indicated stretching frequencies for O–H (3551–3390 cm<sup>-1</sup>), C=O (1745–1650 cm<sup>-1</sup>), and furan ring (1504–1438 cm<sup>-1</sup>) functionalities. The <sup>1</sup>H and <sup>13</sup>C NMR spectra (Tables 1–4) of compounds **1–9** showed the presence of two angular methyl groups at Me-19 ( $\delta_{\text{H}}$  1.06–1.66/ $\delta_{\text{C}}$  19.4–29.7) and Me-20 ( $\delta_{\text{H}}$  0.83–1.35/ $\delta_{\text{C}}$  23.3–29.9), a methoxycarbonyl ( $\delta_{\text{H}}$  3.75–3.87/ $\delta_{\text{C}}$  52.2–53.4), and a  $\gamma$ - or  $\delta$ -lactone at C-8/C-12 or C-18/C-6, respectively. The presence of a furan ring was supported by the signals between  $\delta_{\text{C}}$  125.8–126.8 (C-13), 109.4–109.9 ( $\delta_{\text{H}}$  6.50–6.55, C-14), 144.7–145.0 ( $\delta_{\text{H}}$  7.48–7.50, C-15), and 141.3–141.6 ( $\delta_{\text{H}}$  7.58–7.60, C-16). The configurations of D-glucose and D-xylose, where present, were determined by acid hydrolysis followed by GLC analysis. The A/B ring junction in compounds **1–9** was deduced to be *cis* on the basis of the <sup>13</sup>C NMR chemical shifts of the angular methyl (Me-19), which were found to be in the range  $\delta$  19.4–29.7, while in a *trans* disposition, they would appear at  $\delta$  11–19.<sup>14</sup> The B/C ring fusion was determined on the basis of the <sup>13</sup>C NMR chemical shift for C-11 ( $< \delta$  45.0 for *cis*-fused rings and  $> \delta$  45.0 for *trans*-fused rings).<sup>15</sup>

Compound **1** was obtained as a colorless, gummy solid from the butanolic extract of *T. crispa*. Its HRESIMS showed a [M + H]<sup>+</sup> peak at *m/z* 685.2701, corresponding to the molecular formula C<sub>32</sub>H<sub>45</sub>O<sub>16</sub> (calcd 685.2708). The UV spectrum showed absorptions characteristic for a clerodane-type furanoditerpene. Stretching bands at 3420, 1717, and 1507–1440 cm<sup>-1</sup> in the IR spectrum indicated the presence of OH, ester carbonyl, and aromatic ring functionalities, respectively. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **1** showed the characteristic signals for two angular methyl groups (Me-19 and Me-20), a methoxycarbonyl, and a furan ring. A singlet at  $\delta$  6.40 ( $J_{3,2} = 3.7$  Hz, H-3) was assigned to the C-3 proton of the trisubstituted double bond. The connectivities of the methoxycarbonyl at C-4 and a  $\delta$ -lactone between C-8/C-12 were deduced from the HBMBC correlations between H-3 ( $\delta_{\text{H}}$  6.40, d,  $J_{3,2} = 3.7$  Hz) and C-18 ( $\delta_{\text{C}}$  169.4) and between H-7 ( $\delta_{\text{H}}$  1.47) and C-17 ( $\delta_{\text{C}}$  178.4), respectively. The presence of a  $\delta$ -lactone was also supported by the relatively upfield chemical shift of C-17 ( $\delta_{\text{C}}$  178.4), due to its larger ring size. In  $\gamma$ -lactones, due to angular strain, C-17 resonates above 180 ppm.<sup>16</sup> The HMBC correlation of H-12 ( $\delta_{\text{H}}$  5.53, dd,  $J = 6.0$  Hz,  $J = 5.2$  Hz) and C-13 indicated the substitution of a furan moiety at C-12, while the <sup>13</sup>J cross-peaks of H-12 and C-17/C-9 further supported the presence of a six-membered lactone between C-8 and C-12.<sup>12</sup> The presence of a disaccharide moiety ( $\beta$ -glucopyranosyl-(1 $\rightarrow$ 6)- $\alpha$ -xylopyranose) was inferred by two sets of signals. The first set resonated at  $\delta$  4.42 (d,  $J = 7.7$  Hz, H-1',  $\delta_{\text{C}}$  106.3), 3.22 (t,  $J = 8.3$  Hz, H-2'), 3.37–3.44 (3H, overlap, H-3' to H-5'), 3.89 (dd,  $J = 11.3$  Hz,  $J = 3.7$  Hz,

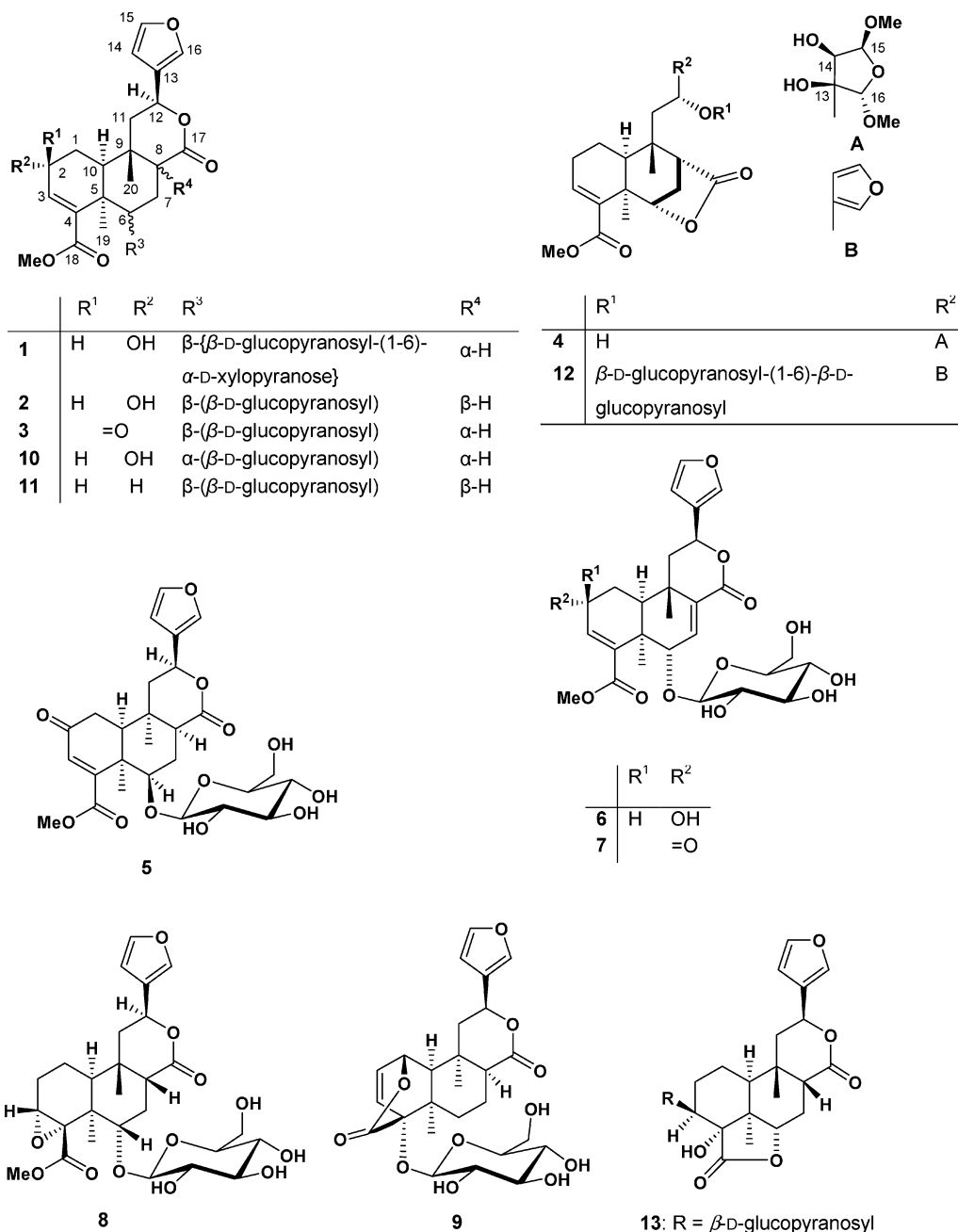
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Chart 1



H-6'a), and 3.67 (dd,  $J = 11.3$  Hz,  $J = 1.4$  Hz, H-6'b), corresponding to a β-glucopyranosyl moiety at C-6. The second set of <sup>1</sup>H NMR saccharide signals appeared at δ 4.78 (d,  $J = 3.6$  Hz, H-1'', δ<sub>C</sub> 100.8), 3.58 (t,  $J = 9.1$  Hz, H-2''), 3.31–3.45 (2H, overlap, H-3'' to H-4''), and 3.52 (m, H-5''), assigned to an α-xylopyranose, connected to β-glucopyranosyl at C-6' through C-1''. This was deduced on the basis of the more downfield oxymethylene signal of C-6' (δ<sub>C</sub> 67.7) in comparison to the signal of C-5'' (δ<sub>C</sub> 63.1).

In the NOESY spectrum of compound **1**, cross-peaks were observed between H-2/Me-20, H-12/H-10, H-8/Me-19, H-10/Me-19, and H-6/H-10, which suggested the orientation of H-2, Me-20, and the furan ring as β and those of H-6, H-8, H-10, H-12, and Me-19 as α. On the basis of these data, the structure of compound **1** was deduced as (2*R*,5*R*,6*R*,8*S*,9*S*,10*S*,12*S*)-15,16-epoxy-2-hydroxy-6-*O*-{β-D-glucopyranosyl-(1→6)-α-D-xylopyranosyl}-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester.

Compound **2** was obtained from the butanolic extract of the plant as a colorless, gummy solid using recycling HPLC (GS-320). Its

molecular formula was deduced as C<sub>27</sub>H<sub>37</sub>O<sub>12</sub> on the basis of the molecular ion peak at  $m/z$  553.2215, corresponding to the [M + H]<sup>+</sup> peak in the HRESIMS. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **2** showed signals similar to those of compound **1**, except for the absence of signals for the xylose moiety. Along with this, differences were found in the chemical shifts of H-6 (δ<sub>H</sub> 4.53, δ<sub>C</sub> 78.4), H-10 (δ<sub>H</sub> 2.56, δ<sub>C</sub> 42.0), H-12 (δ<sub>H</sub> 5.77, δ<sub>C</sub> 71.3), and Me-20 (δ<sub>H</sub> 1.09, δ<sub>C</sub> 29.4). This indicated a possible difference in the relative configuration of compound **2**. The relative configurations at C-2, C-5, C-6, C-8, C-9, C-10, and C-12 were determined from the NOESY spectrum, in which correlations were seen between H-2/Me-20, H-8/Me-20, H-12/H-10, H-10/Me-19, and H-6/H-10, indicating that H-2, Me-20, and H-8 are β-oriented, while H-6, H-10, H-12, and Me-19 are α-oriented. On the basis of these arguments, the compound was deduced as (2*R*,5*R*,6*R*,8*R*,9*S*,10*S*,12*S*)-15,16-epoxy-2-hydroxy-6-*O*-(β-D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester.

Compound **3** was obtained from the butanolic extract of *T. crispata*. Its molecular formula was deduced as C<sub>27</sub>H<sub>34</sub>O<sub>12</sub> on the basis of

**Table 1.** <sup>1</sup>H NMR Data (CD<sub>3</sub>OD) of Compounds 1–3

position	1 <sup>a</sup>	2 <sup>a</sup>	3 <sup>b</sup>
	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{H}}$ (J in Hz)
1	2.29, m 1.99, m	2.25, m 2.00, m	2.93, dd (18.9, 6.5) 2.55, d (18.9)
2	4.51, td (8.3, 3.7)	4.49, td (8.3, 3.8)	
3	6.40, d (3.7)	6.59, d (3.8)	6.26, s
6	4.57, brd (3.0)	4.53, brd (4.6)	4.39, brd (2.6)
7	2.27, m 1.47, t (13.1)	2.67, ddd (14.0, 4.6, 1.5) 1.73, dd (14.0, 6.5)	2.39, dt (14.4, 3.2) 1.56, dd (14.4, 11.7)
8	2.25, m	2.33, dd (6.5, 1.5)	3.48, dd (11.7, 2.2)
10	3.45, m	2.56, brd (6.02)	2.65, d (6.5)
11	2.00, m 2.03, m	2.20, dd (14.6, 12.4) 1.65 dd (14.6, 1.5)	2.10, dd (14.0, 6.2) 1.95, dd (14.0, 11.1)
12	5.53, dd (6.0, 5.2)	5.77, brd (12.4)	5.53, dd (11.1, 6.2)
14	6.51, brs	6.53, brs	6.50, d (1.0)
15	7.48, brs	7.48, dd (1.5, 1.0)	7.50, t (1.5)
16	7.60, s	7.60, brs	7.58, brs
19	1.55 s	1.55, s	1.65, s
20	0.96, s	1.09, s	0.83, s
MeOOC–	3.75, s	3.75, s	3.85, s
1'	4.42, d (7.7)	4.41, d (7.5)	4.44, d (7.6)
2'	3.22, t (8.3) <sup>c</sup>	3.17, dd (7.9, 7.5)	3.27, m
3'	3.37, m	3.30, m <sup>c</sup>	3.37, m <sup>c</sup>
4'	3.39, m	3.39, dd (9.4, 9.2)	3.37, m
5'	3.44, m	3.20, m <sup>c</sup>	3.29, m <sup>c</sup>
6'	3.89, dd (11.3, 3.7) 3.67, dd (11.3, 1.4)	3.80, dd (11.9, 2.5) 3.70, dd (11.9, 3.9)	3.80, dd (11.9, 2.3) 3.70, dd (11.9, 4.2)
1''	4.78, d (3.6)		
2''	3.58, t (9.0) <sup>c</sup>		
3''	3.31, m		
4''	3.45, m		
5''	3.52, m		

<sup>a</sup> Measured at 500 MHz. <sup>b</sup> Measured at 300 MHz. <sup>c</sup> Signals are interchangeable in each column.

**Table 2.** <sup>1</sup>H NMR Data (CD<sub>3</sub>OD) of Compounds 5, 8, and 9

position	5 <sup>a</sup>	8 <sup>b</sup>	9 <sup>b</sup>
	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{H}}$ (J in Hz)	$\delta_{\text{H}}$ (J in Hz)
1	2.89, dd (16.8, 6.6) 2.61, dd (16.8, 5.3)	1.70, m 1.57, m	5.30, brd (4.6)
2		2.04, m 1.90, m	6.61, dd (7.2, 4.6)
3	6.42, s	3.35, m	6.81, brd (7.2)
6	4.43, d (3.4)	4.58, dd (9.5, 9.3)	1.94, m 1.49, m
7	2.80, dd (14.3, 3.4) 1.79, dd (14.3, 6.4)	2.62, m 2.59, m	2.55, m 2.08 m
8	2.40, dd (6.4, 1.5)	2.60, m	2.57, m
10	2.92, m	1.70, brs	1.81, brs
11	2.20, dd (14.8, 1.5) 1.68, dd (14.8, 12.1)	2.22, dd (14.8, 1.8) 1.80, dd (14.8, 11.8)	2.39, dd (14.7, 4.1) 1.98, m
12	5.78, dd (12.1, 1.5)	5.50, brd (11.2)	5.56, dd (12.2, 4.1)
14	6.54, d (1.0)	6.52, d (1.0)	6.55, d (1.0)
15	7.49, t (1.3)	7.50, t (1.5)	7.50, brs
16	7.60, brs	7.59, brs	7.59, brs
19	1.66, s	1.30, s	1.06, s
20	0.98, s	1.35, s	1.20, s
MeOOC–	3.85, s	3.78, s	
1'	4.42, d (7.7)	4.34, d (7.6)	4.71, d (6.1)
2'	3.18, dd (8.9, 7.7)	3.27, m	3.38, m
3'	3.31, m <sup>c</sup>	3.37, m <sup>c</sup>	3.38, m <sup>c</sup>
4'	3.38, t (9.3)	3.37, m	3.38, m
5'	3.22, m <sup>c</sup>	3.29, m <sup>c</sup>	3.29, m <sup>c</sup>
6'	3.81, dd (11.8, 1.9) 3.70, dd (11.8, 4.5)	3.80, dd (11.8, 1.9) 3.70, dd (11.8, 4.5)	3.81, dd (11.9, 2.1) 3.65, dd (11.9, 5.2)

<sup>a</sup> Measured at 600 MHz. <sup>b</sup> Measured at 300 MHz. <sup>c</sup> Signals are interchangeable in each column.

the peak at  $m/z$  551.2133, corresponding to  $[M + H]^+$  in the HRESIMS. The <sup>1</sup>H and <sup>13</sup>C NMR spectra (Tables 1 and 3) displayed singlets for the two methyl groups (Me-19 and Me-20), a methoxycarbonyl, and a furan ring characteristic of a furanoditerpene. A detailed analysis of the NMR data for compound 3 showed it to be similar to compound 2, namely, a furanoditerpene  $\delta$ -lactone glucosylated at C-6 (Tables 1 and 3). However, the C-2 position in compound 3 was oxidized as was inferred from the presence of

an additional carbonyl group at  $\delta_{\text{C}}$  200.3. The HMBC correlations of H-1 ( $\delta_{\text{H}}$  2.55), H-10 ( $\delta_{\text{H}}$  2.65), and H-3 ( $\delta_{\text{H}}$  6.26) with C-2 further supported the keto functionality in compound 3. The relative configuration in compound 3 was deduced on the basis of NOESY correlations between H-6/Me-19, H-8/H-10, H-8/H-6, H-10/Me-19, and H-12/H-10. These showed Me-20, the  $\beta$ -glucopyranosyl at C-6, and the furan ring at C-12 to be  $\beta$ -oriented, with H-8, H-10, H-12, and Me-19 all  $\alpha$ -oriented. On the basis of the above discussion, compound was assigned as (5*R*,6*R*,8*S*,9*R*,10*R*,12*S*)-15,16-epoxy-2-oxo-6-*O*-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester.

Compound 4 gave the molecular formula C<sub>27</sub>H<sub>34</sub>O<sub>10</sub>, as deduced from its positive-ion HRESIMS at  $m/z$  471.2312  $[M + H]^+$ . The IR spectrum showed the presence of a  $\gamma$ -lactone carbonyl at 1745 cm<sup>-1</sup>. The <sup>1</sup>H and <sup>13</sup>C NMR spectra exhibited a downfield oxymethine signal at  $\delta_{\text{H}}$  5.57 ( $\delta_{\text{C}}$  84.1) for H-6. The upfield oxymethine signal at  $\delta_{\text{H}}$  3.84 ( $\delta_{\text{C}}$  69.0) was assigned to H-12. The HMBC correlations of H-6 ( $\delta_{\text{H}}$  5.57), H-7 ( $\delta_{\text{H}}$  2.27 and 2.18), and H-8 ( $\delta_{\text{C}}$  2.43) with C-17 ( $\delta_{\text{C}}$  180.0, C=O) indicated the presence of a  $\gamma$ -lactone. This was also supported by the downfield chemical shift of C-17 ( $\delta_{\text{C}}$  180.5) due to ring strain.<sup>16</sup> The presence of a tetrahydrofuran moiety was inferred from the signals at  $\delta$  82.0, 76.2, 111.7, and 109.5, corresponding to C-13, C-14, C-15, and C-16, respectively.<sup>9</sup> The long-range <sup>3</sup>J correlations of H-12 ( $\delta_{\text{H}}$  3.84) with C-9 ( $\delta_{\text{C}}$  39.1) and C-13 ( $\delta_{\text{C}}$  82.0) along with its COSY correlations with the H-11 methylene protons ( $\delta_{\text{H}}$  1.69 and 1.59) indicated a linkage between ring B and the tetrahydrofuran moiety through C-11 and C-12. The <sup>1</sup>H NMR spectrum of compound 4 also showed two methoxy singlets at  $\delta_{\text{H}}$  3.47 ( $\delta_{\text{C}}$  56.4) and 3.41 ( $\delta_{\text{C}}$  54.8). The location of these methoxy substituents at C-15 ( $\delta_{\text{C}}$  111.7) and C-16 ( $\delta_{\text{C}}$  109.5) was deduced from HMBC correlations (Figure 1). On the basis of a comparison with the literature data for known related compounds,<sup>9</sup> it was deduced that C-13 and C-14 are hydroxylated in compound 4. The relative configuration at various centers in compound 4 was deduced from NOESY and ROESY experiments (Figure 1). Cross-peaks were observed between H-10/Me-19, H-8/Me-20, H-12/Me-20, H-8/H-16, H-12/H-16, and H-14/H-15, indicating that H-10, Me-19, the  $\gamma$ -lactone ring, H-14, and H-15 are  $\alpha$ -oriented, while H-8, H-12, Me-20, the methylene bridge at C-7, the C-13 hydroxy group, and H-16 are all  $\beta$ -oriented (Figure 1). On the basis of these data, the structure of 7 was identified as methyl (2*R*,7*S*,8*S*)-8-[(2*S*)-2-(3,4-dihydroxy-2,5-dimethoxytetrahydro-3-furanyl)-2-hydroxyethyl]-2,8-dimethyl-10-oxo-11-oxatricyclo[7.2.1.0<sup>2,7</sup>]dodec-3-ene-3-carboxylate and named, trivially, rumphiol E.

Compound 5 displayed an  $[M + H]^+$  peak at  $m/z$  551.2171 in the HRESIMS, corresponding to the molecular formula C<sub>27</sub>H<sub>35</sub>O<sub>12</sub>. The <sup>1</sup>H and <sup>13</sup>C NMR data of 5 were similar to compound 3, except for slight differences in chemical shifts and splitting patterns, with differences found for C-8 ( $\delta_{\text{H}}$  2.40,  $\delta_{\text{C}}$  45.4), C-11 ( $\delta_{\text{H}}$  2.20, 1.18,  $\delta_{\text{C}}$  42.4), and C-20 ( $\delta_{\text{H}}$  0.98,  $\delta_{\text{C}}$  27.9). In addition to these, H-12 in compound 5 appeared more deshielded, resonating at  $\delta_{\text{H}}$  5.78. The NOESY spectrum of compound 5 showed correlations between H-6/Me-19, H-6/H-8, H-10/Me-19, H-8/Me-20, and H-12/H-10. These indicated that the  $\beta$ -glucopyranosyl at C-6 and the furan ring at C-12 are  $\beta$ -oriented, while H-6, H-8, H-10, H-12, Me-19, and Me-20 are  $\alpha$ -oriented. Therefore, it was concluded that compound 5 is a C-9-epimer of compound 3. Compound 5 was deduced to have the structure (5*R*,6*R*,8*S*,9*R*,10*S*,12*S*)-15,16-epoxy-2-oxo-6-*O*-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester.

Compound 6 was isolated as a colorless gum. Its HRESIMS showed a  $[M + Na]^+$  peak at  $m/z$  573.1980, corresponding to the molecular formula C<sub>27</sub>H<sub>34</sub>O<sub>12</sub>Na. The glycosidic nature of 6 was inferred from the EIMS aglycon peak at  $m/z$  388 due to the loss of 162 amu, corresponding to a hexose sugar unit. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of 6 showed the presence of two angular methyl

**Table 3.**  $^{13}\text{C}$  NMR Data ( $\text{CD}_3\text{OD}$ ) of Compounds **1–3**, **5**, **8**, and **9**

position	<b>1<sup>a</sup></b>	<b>2<sup>b</sup></b>	<b>3<sup>a</sup></b>	<b>5<sup>c</sup></b>	<b>8<sup>a</sup></b>	<b>9<sup>c</sup></b>
	$\delta_{\text{C}}$ mult.	$\delta_{\text{C}}$ mult.	$\delta_{\text{C}}$ mult.	$\delta_{\text{C}}$ mult.	$\delta_{\text{C}}$ mult.	$\delta_{\text{C}}$ mult.
1	27.3, CH <sub>2</sub>	29.1, CH <sub>2</sub>	36.2, CH <sub>2</sub>	36.2, CH <sub>2</sub>	16.3, CH <sub>2</sub>	74.8, CH
2	64.6, CH	64.9, CH	200.3, qC	200.8, qC	22.3, CH <sub>2</sub>	131.7, CH
3	140.7, CH	141.7, CH	133.4, CH	134.2, CH	60.3, CH	132.5, CH
4	140.3, qC	139.7, qC	156.7, qC	155.7, qC	63.2, qC	87.5, qC
5	42.7, qC	42.0, qC	44.0, qC	43.4, qC	40.9, qC	40.0, qC
6	80.8, CH	78.4, CH	79.6, CH	77.9, CH	78.5, CH	27.1, CH <sub>2</sub>
7	29.3, CH <sub>2</sub>	29.7, CH <sub>2</sub>	28.0, CH <sub>2</sub>	29.9, CH <sub>2</sub>	30.6, CH <sub>2</sub>	18.5, CH <sub>2</sub>
8	51.7, CH	46.6, CH	49.8, CH	45.4, CH	49.3, CH	45.2, CH
9	38.3, qC	35.9, qC	38.5, qC	36.4, qC	35.6, qC	36.2, qC
10	42.1, CH	42.0, CH	41.4, CH	40.6, CH	39.9, CH	47.4, CH
11	46.2, CH <sub>2</sub>	43.2, CH <sub>2</sub>	45.2, CH <sub>2</sub>	42.4, CH <sub>2</sub>	42.0, CH <sub>2</sub>	42.3, CH <sub>2</sub>
12	71.9, CH	71.3, CH	71.7, CH	71.3, CH	71.3, CH	72.7, CH
13	125.9, qC	126.9, qC	125.8, qC	126.8, qC	126.6, qC	126.6, qC
14	109.9, CH	109.6, CH	109.8, CH	109.6, CH	109.4, CH	109.6, CH
15	144.9, CH	144.7, CH	145.0, CH	144.8, CH	144.9, CH	145.0, CH
16	141.6, CH	141.3, CH	141.5, CH	141.3, CH	141.3, CH	141.4, CH
17	178.4, qC	177.2, qC	177.5, qC	176.4, qC	176.9, qC	176.7, qC
18	169.4, qC	169.1, qC	168.3, qC	168.0, qC	172.2, qC	175.2, qC
19	29.7, CH <sub>3</sub>	28.7, CH <sub>3</sub>	28.3, CH <sub>3</sub>	27.4, CH <sub>3</sub>	19.4, CH <sub>3</sub>	24.6, CH <sub>3</sub>
20	23.8, CH <sub>3</sub>	29.4, CH <sub>3</sub>	23.3, CH <sub>3</sub>	27.9, CH <sub>3</sub>	29.9, CH <sub>3</sub>	28.1, CH <sub>3</sub>
MeOOC–	52.5, CH <sub>3</sub>	52.4, CH <sub>3</sub>	52.2, CH <sub>3</sub>	53.2, CH <sub>3</sub>	53.4, CH <sub>3</sub>	
1'	106.3, CH	105.8, CH	106.2, CH	105.9, CH	104.9, CH	101.5, CH
2'	75.5, CH <sup>d</sup>	75.4, CH	75.6, CH	75.4, CH	75.4, CH	75.1, CH
3'	78.1, CH	78.0, CH <sup>d</sup>	78.1, CH <sup>d</sup>	77.9, CH <sup>d</sup>	78.0, CH <sup>d</sup>	78.3, CH <sup>d</sup>
4'	71.2, CH	70.6, CH	71.3, CH	70.6, CH	71.7, CH	71.2, CH
5'	76.5, CH <sup>d</sup>	77.2, CH <sup>d</sup>	77.8, CH <sup>d</sup>	77.3, CH <sup>d</sup>	77.8, CH <sup>d</sup>	78.0, CH <sup>d</sup>
6'	67.7, CH <sub>2</sub>	62.3, CH <sub>2</sub>	62.3, CH <sub>2</sub>	62.3, CH <sub>2</sub>	62.6, CH <sub>2</sub>	62.5, CH <sub>2</sub>
1''	100.8, CH					
2''	75.6, CH <sup>d</sup>					
3''	74.0, CH					
4''	71.6, CH					
5''	63.1, CH <sub>2</sub>					

<sup>a</sup> Measured at 125 MHz. <sup>b</sup> Measured at 100 MHz. <sup>c</sup> Measured at 150 MHz. <sup>d</sup> Signals are interchangeable in each column.

groups, a methoxycarbonyl, and a furan ring. The proton signals resonating at  $\delta_{\text{H}}$  6.40 ( $\delta_{\text{C}}$  138.5, C-3;  $\delta_{\text{C}}$  141.5, C-4) and 6.94 ( $\delta_{\text{C}}$  140.9, C-7;  $\delta_{\text{C}}$  137.0, C-8) were assigned to two trisubstituted double bonds. The HBMBC correlations of H-3 ( $\delta_{\text{H}}$  6.40) with C-18 and of H-7 ( $\delta_{\text{H}}$  6.94) with C-17 suggested the occurrence of a methoxycarbonyl at C-4 and a  $\delta$ -lactone between C-8 and C-12. The HMBC correlations of H-12 ( $\delta_{\text{H}}$  5.24) and C-13 indicated the presence of a furan moiety at C-12. The spectroscopic data of **6** were closely related to those of the known compound borapetoside F,<sup>11</sup> with an additional hydroxymethine signal at  $\delta_{\text{H}}$  4.51 ( $\delta_{\text{C}}$  64.0) assigned to the oxygenated methine carbon at C-2. The presence of a  $\beta$ -glucopyranosyl moiety was inferred from the NMR signals at  $\delta_{\text{C}}$  105.7 ( $\delta_{\text{H}}$  4.41,  $J = 7.6$  Hz) and supported by the MS fragment ion at  $m/z$  389 in the FABMS, due to the loss of 162 amu. The relative configuration in compound **6** (Figure 2) was deduced on the basis of a ROESY experiment. The correlations obtained showed that H-10, Me-19, H-12, the hydroxy group at C-2, and the  $\beta$ -glucopyranosyl moiety at C-6 are all  $\alpha$ -oriented, while Me-20, H-6, H-2, and the furan ring at C-12 are  $\beta$ -oriented (Figure 1). On the basis of these data, the structure of **6** was deduced as (2*R*,5*R*,6*S*,9*S*,10*S*,12*S*)-15,16-epoxy-2-hydroxy-6-*O*-( $\beta$ -D-glucopyranosyl)-cleroda-3,7,13(16),14-tetraen-17,12-olid-18-oic acid methyl ester.

Compound **7** has the molecular formula  $\text{C}_{27}\text{H}_{33}\text{O}_{12}$ , as deduced from its HRESIMS, with the  $[\text{M} + \text{H}]^+$  peak at  $m/z$  549.1976. Its spectroscopic data suggested structural features similar to compound **6**, except for the oxygenation at C-2, as inferred from an additional signal at  $\delta_{\text{C}}$  200.0. The  $^{13}\text{C}$  NMR spectrum of compound **7** showed signals for two quaternary carbons (C-5 and C-9), two secondary carbons (C-1 and C-11), two oxygenated methine groups (C-6 and C-12), and two trisubstituted double bonds (C-3/C-4 and C-7/C-8). The relative configuration in compound **7** was deduced on the basis of NOESY correlations, which were observed between H-10/Me-19, H-12/Me-19, and Me-20/H-6, and indicated that H-10, H-12,

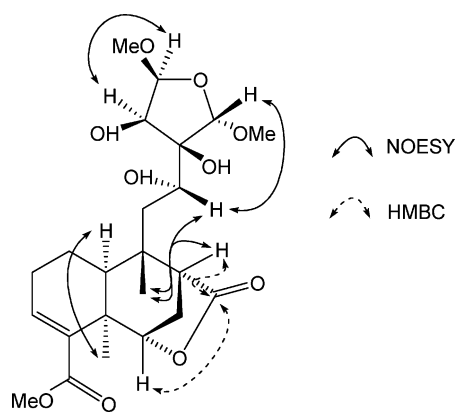
and Me-19 all are  $\beta$ -oriented, while H-6 and Me-20 are  $\alpha$ -oriented. On the basis of these data, the new compound **7** was assigned the structure (5*R*,6*S*,9*S*,10*S*,12*S*)-15,16-epoxy-2-oxo-6-*O*-( $\beta$ -D-glucopyranosyl)-cleroda-3,7,13(16),14-tetraen-17,12-olid-18-oic acid methyl ester.

Compound **8** displayed a  $[\text{M} + \text{H}]^+$  peak at  $m/z$  553.2255 in the HRESIMS, corresponding to the molecular formula  $\text{C}_{27}\text{H}_{37}\text{O}_{12}$ . The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of compound **8** showed characteristic signals for a clerodane-type furanoditerpenoid [ $\delta_{\text{H}}$  1.30 ( $\delta_{\text{C}}$  19.4), 1.35 ( $\delta_{\text{C}}$  29.9), and 3.78 ( $\delta_{\text{C}}$  53.4) assigned to Me-19, Me-20, and  $\text{CO}_2\text{Me}$ -18, respectively]. The signals for the furan moiety appeared at  $\delta_{\text{H}}$  6.52 ( $\delta_{\text{C}}$  109.4), 7.50 ( $\delta_{\text{C}}$  144.9), and 7.59 ( $\delta_{\text{C}}$  141.3) in addition to the  $^{13}\text{C}$  NMR signal at  $\delta_{\text{C}}$  126.6 (C-13). The resonances at  $\delta$  4.34 (d,  $J = 7.6$  Hz, H-1'), 3.80 (dd,  $J = 11.8$  Hz,  $J = 1.9$  Hz, H-6'a), and 3.70 (dd,  $J = 11.8$  Hz,  $J = 4.5$  Hz, H-6'b) were assigned to a  $\beta$ -glucopyranosyl moiety at C-6 ( $\delta_{\text{C}}$  78.5) of the clerodane-type aglycon skeleton. In turn, the signals at  $\delta$  3.25 (overlap, H-3) were assigned to the proton of an epoxide ring between C-3 and C-4 ( $\delta_{\text{C}}$  60.3/ $\delta_{\text{C}}$  63.2). The HMBC spectrum showed correlations of H-3 with C-1, C-2, and the epoxy-substituted methine carbon C-3 ( $\delta_{\text{C}}$  60.3). The protons of the ester methoxy at  $\delta_{\text{H}}$  3.78 exhibited HMBC correlations with quaternary carbons at C-4 ( $\delta_{\text{C}}$  63.2), C-5 ( $\delta_{\text{C}}$  40.9), C-18 ( $\delta_{\text{C}}$  172.2), and C-3 ( $\delta_{\text{C}}$  60.3). Furthermore, the methine signal at  $\delta_{\text{H}}$  4.58 (H-6) gave correlations with C-5, C-6, and C-7 ( $\delta_{\text{C}}$  40.9, 78.5, and 30.6, respectively). The relative configuration in compound **8** was determined on the basis of the NOESY spectrum, which showed cross-peaks between H-3/H-6, H-6/H-8, H-6/Me-20, H-8/Me-20, H-10/Me-19, and H-12/H-10, thus indicating that H-3, the methoxycarbonyl group at C-4, Me-20, H-6, H-8, and the furan ring at C-12 are all  $\beta$ -oriented, while the epoxide ring (C-3/C-4), Me-19, the D-glucopyranosyl, and H-12 are  $\alpha$ -oriented. On the basis of these observations, the structure of compound was proposed as (3*R*,4*R*,5*R*,6*S*,8*R*,9*S*,10*S*,12*S*)-15,16-

**Table 4.** NMR Data of Compounds **4**, **6**, and **7** (**4** in CDCl<sub>3</sub>; **6**, **7** in CD<sub>3</sub>OD)

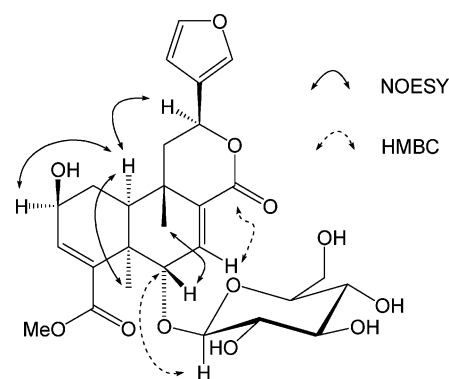
position	<b>4</b>		<b>6</b>		<b>7</b>	
	$\delta_C$ mult. <sup>a</sup>	$\delta_H$ (J in Hz) <sup>d</sup>	$\delta_C$ mult. <sup>b</sup>	$\delta_H$ (J in Hz) <sup>d</sup>	$\delta_C$ mult. <sup>b</sup>	$\delta_H$ (J in Hz) <sup>d</sup>
1	16.3, CH <sub>2</sub>	1.96, m 1.94, m	30.8, CH <sub>2</sub>	2.09, m 1.93, m	36.5, CH <sub>2</sub>	2.67, m
2	24.0, CH <sub>2</sub>	2.40, brd (5.4) 2.30, m	64.0, CH	4.35, m	200.0, qC	
3	142.8, CH	7.03, dd (3.8, 3.6)	138.5, CH	6.40, d (3.5)	133.4, CH	6.31, s
4	133.8, qC		141.5, qC		156.3, qC	
5	38.9 <sup>e</sup> , qC		43.5, qC		43.0, qC	
6	84.1, CH	5.57, brd (6.0)	81.1, CH	4.52, d (2.4)	75.6, CH	4.77, m
7	29.4, CH <sub>2</sub>	2.27, m 2.18, m	140.9, CH	6.94, d (2.4)	136.2, CH	6.77, d (4.4)
8	48.2, CH	2.43, brd (5.4)	137.0, qC		138.6, qC	
9	39.1 <sup>e</sup> , qC		38.2, qC		38.2, qC	
10	44.3, CH	1.41, dd (5.2, 2.1)	46.5, CH	2.06, m	46.7, CH	2.63, m
11	39.7, CH <sub>2</sub>	1.69, dd (15.5, 7.5) 1.56, m	45.1, CH <sub>2</sub>	2.40, dd (14.6, 3.3) 2.04, m	45.6, CH <sub>2</sub>	2.22, dd (14.3, 3.7) 2.06, dd (14.3, 11.7)
12	69.0, CH	3.84, brd (7.5)	72.4, CH	5.24, dd (11.4, 3.3)	72.9, CH	5.21, dd (11.7, 3.7)
13	82.0, qC		125.3, qC		125.1, qC	
14	76.2, CH	4.23, brs	109.7, CH	6.53, d (1.0)	109.7, CH	6.51, brs
15	111.7, CH	4.97, d (2.7)	145.0, CH	7.51, t (1.5)	145.0, CH	7.50, t (1.4)
16	109.5, CH	4.90, s	141.5, CH	7.61, brs	141.5, CH	7.59, s
17	180.5, qC		171.2, qC		172.4, qC	
18	166.7, qC		172.0, qC		168.6, qC	
19	27.0, CH <sub>3</sub>	1.32, s	23.9, CH <sub>3</sub>	1.45, s	27.1, CH <sub>3</sub>	1.69, s
20	19.9, CH <sub>3</sub>	1.16, s	27.4, CH <sub>3</sub>	1.12, s	27.4, CH <sub>3</sub>	1.03, s
MeOOC-	51.6, CH <sub>3</sub>	3.71, s	52.3, CH <sub>3</sub>	3.76, s	53.3, CH <sub>3</sub>	3.86, s
MeO-15	56.4, CH <sub>3</sub>	3.47, s				
MeO-16	54.8, CH <sub>3</sub>	3.41, s				
1'			105.7, CH	4.41, d (7.6)	105.8, CH	4.51, d (7.5)
2'			75.5, CH	3.17, dd (7.5, 7.9)	75.7, CH	3.23, t (7.7)
3'			77.5, CH <sup>e</sup>	3.30, m	78.0, CH <sup>e</sup>	3.33, m
4'			71.3, CH	3.39, dd (9.4, 9.2)	71.5, CH	3.32, m
5'			78.1, CH <sup>e</sup>	3.20, m	78.1, CH <sup>e</sup>	3.33, m
6'			62.5, CH <sub>2</sub>	3.85, dd (12.0, 2.0) 3.70, dd (12.0, 4.8)	62.6, CH <sub>2</sub>	3.85, dd (11.7, 1.1) 3.70, dd (11.7, 4.0)

<sup>a</sup> Measured at 75 MHz. <sup>b</sup> Measured at 100 MHz. <sup>c</sup> Measured at 300 MHz. <sup>d</sup> Measured at 500 MHz. <sup>e</sup> Signals are interchangeable in each column.

**Figure 1.** Key NOESY and HMBC interactions of compound **4**.

epoxy-3,4-epoxy-6-*O*-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester.

The positive-ion HRESIMS of compound **9** showed a  $[M + H]^+$  peak at  $m/z$  521.2043, corresponding to the molecular formula C<sub>26</sub>H<sub>34</sub>O<sub>11</sub>. In the <sup>1</sup>H NMR spectrum, a broad doublet at  $\delta_H$  5.30 ( $\delta_C$  74.8) was due to coupling of H-1 and H-2 ( $\delta_H$  6.61). Protons of a disubstituted olefin (C-2/C-3) appeared as a double doublet at  $\delta_H$  6.61 ( $\delta_C$  131.7, C-2) and a doublet at  $\delta_H$  6.81 ( $\delta_C$  132.5, C-3). In the <sup>13</sup>C NMR spectrum, the quaternary carbons resonating at  $\delta_C$  87.5, 176.7, and 175.2 were assigned to C-4, C-17, and C-18, respectively. The <sup>1</sup>H and <sup>13</sup>C NMR spectra also showed signals for a monosubstituted furan moiety at C-12 ( $\delta_C$  72.7), a  $\beta$ -glucopyranosyl unit at C-4, and two angular methyls at C-5 ( $\delta_C$  40.0) and C-9 ( $\delta_C$  36.2).<sup>10–12</sup> The HMBC spectrum showed correlations of the methine carbon ( $\delta$  74.8) with protons at  $\delta_H$  6.61 (H-2), 6.81

**Figure 2.** Key NOESY and HMBC interactions of compound **6**.

(H-3), and 1.81 (H-10). The connectivity of the  $\beta$ -glucopyranosyl unit at C-4 was inferred from the correlation of the anomeric proton (H-1',  $\delta_H$  4.71) with the quaternary carbon at C-4 ( $\delta_C$  87.5). The relative configuration of compound **9** was determined on the basis of NOESY interactions. The correlations between H-1/H-10, H-10/Me-19, H-8/Me-20, H-8/H-10, and H-12/H-10 indicated that H-1, H-10, Me-19, H-8, Me-20, H-12, and  $\beta$ -D-glucopyranosyl are all  $\alpha$ -oriented, while the lactone ring at C-1/C-4 and furan ring at C-12 are  $\beta$ -oriented. Therefore, compound **9** was deduced to be (1*R*,4*S*,5*R*,8*S*,9*R*,10*S*,12*S*)-15,16-epoxy-4-*O*-( $\beta$ -D-glucopyranosyl)-cleroda-2,13(16),14-triene-17(12),18(1)-diolide.

Analysis of the <sup>13</sup>C NMR spectrum of borapetoside A (**13**) (Table 5) revealed that certain assignments were different from those reported, particularly the chemical shifts of Me-19 and Me-20.<sup>17</sup> This suggested a different configuration at the A/B ring junction. In order to resolve this ambiguity, the 2D NMR data of **13** were

**Table 5.** NMR Data of Compound **13** (CD<sub>3</sub>OD)

position	$\delta_C$ mult. <sup>a</sup>	$\delta_H$ (J in Hz) <sup>b</sup>	HMBC
1	19.6, CH <sub>2</sub>	1.73, m, 1.61, m	1, 2, 3, 9, 10, 19
2	28.3, CH <sub>2</sub>	2.1, m, 1.70, m	1, 3, 4, 10, 18
3	80.0, CH	4.01, brs	1, 1', 2, 3, 4, 5, 18
4	81.6, qC		
5	47.3, qC		
6	76.6, CH	5.0, dd (11.8, 4.4)	3, 5, 6, 7, 8, 19
7	26.3, CH <sub>2</sub>	2.31, m, 2.22 m	5, 6, 7, 8, 9, 17, 20
8	48.1, CH <sup>c</sup>	2.74, dd (12.4, 5.6)	6, 7, 9, 10, 11, 12, 17, 20
9	36.1, qC		
10	48.3, CH <sup>c</sup>	1.87, m	1, 2, 5, 9, 10, 11, 19
11	45.1, CH <sub>2</sub>	2.37, m, 1.98, m	9, 10, 11, 12, 13, 17, 20
12	72.3, CH	5.87, dd (11.8, 4.4)	9, 11, 12, 13, 14, 16, 17
13	126.1, qC		
14	109.5, CH	6.52, d (1.0)	12, 13, 14, 15, 16
15	145.1, CH	7.50, dd (1.6, 1.6)	13, 14, 15, 16
16	141.6, CH	7.62, brs	13, 14, 15, 16
17	175.9, qC		
18	180.0, qC		
19	18.1, CH <sub>3</sub>	1.21, s	4, 5, 6, 10, 18, 19
20	33.3, CH <sub>3</sub>	1.10, s	1, 8, 9, 10, 11
1'	105.0, CH	4.30, d (7.7)	1', 3, 5'
2'	75.1, CH	3.14, m	
3'	77.7, CH <sup>c</sup>	3.32, m	
4'	71.4, CH	3.35, m	
5'	78.0, CH <sup>c</sup>	3.23, m	
6'	62.0, CH <sub>2</sub>	3.81, dd (11.9, 2.0) 3.67, dd (11.9, 5.1)	

<sup>a</sup> Measured at 100 MHz. <sup>b</sup> Measured at 300 MHz. <sup>c</sup> Signals are interchangeable in each column.

studied carefully. The HMQC and HMBC spectra supported the structure reported previously,<sup>17</sup> but the assignments of Me-19 and Me-20 need to be revised. The proton signal at  $\delta_H$  1.21 ( $\delta_C$  18.2) was assigned to Me-20, and  $\delta_H$  1.10 ( $\delta_C$  33.3) was due to Me-20 on the basis of HMQC and HMBC correlations (Table 5).

Compounds **10–13** were assayed for cytotoxicity against cancer PC-3 (human prostate) and normal 3T3 (mouse fibroblast) cell lines. The 50% inhibitory concentrations (IC<sub>50</sub>) in the case of both cell lines were >10  $\mu$ M. These results are in accordance with the reported cytotoxic activity of crude dichloromethane extract of certain *Tinospora* species, for which no cytotoxic effects have been observed.<sup>1–4</sup>

## Experimental Section

**General Experimental Procedures.** Optical rotations were measured on a JASCO DIP-360 digital polarimeter in methanol, unless otherwise stated. UV spectra were recorded on a Shimadzu UV240 spectrophotometer in MeOH solution, and IR spectra were recorded as KBr disks on a JASCO A-302 spectrophotometer. 1D NMR spectra were recorded on a Bruker Avance spectrometer, operating at either 400 (<sup>1</sup>H NMR) or 100 (<sup>13</sup>C NMR) MHz, unless otherwise stated. The chemical shifts values are given in  $\delta$  (ppm), referenced to the residual solvent signal (CD<sub>3</sub>OD/CDCl<sub>3</sub>), while coupling constants (*J*) were measured in Hz. 2D NMR spectra were taken on a Bruker AMX 500 NMR spectrometer. Electron-impact mass spectra (EIMS) were taken at 70 eV on a Finnigan MAT-112 or MAT-312 instrument. Fast-atom bombardment mass spectra (FABMS) were measured as a glycerol matrix on a JEOL HX-110 mass spectrometer. TLC was performed on precoated silica gel plates (DC-Alugram 60 UV254 of E. Merck), and the spots were observed first under UV light (254 nm) and then stained with ceric(IV) sulfate spray reagent and heated until the appearance of a color. Diaion HP-20 (Mitsubishi Chemical Industries, Tokyo, Japan), ODS C<sub>18</sub> (63–212  $\mu$ m, Wako Pure Chemical Industries, Ltd., Japan), polyamide-6 DF (Riedel-De Haen AG), and silica gel (E. Merck, 230–400  $\mu$ m mesh) were used as adsorbents. Recycling preparative HPLC separation was performed on a JAI LC-908W instrument (Japan Analytical Industry) with YMC ODS H-80, L-80, and GS-320 columns (YMC, Ltd., Kyoto, Japan).

**Plant Material.** *Tinospora crispa* was collected from the herbal garden of the Laboratory of Natural Products (LHS), University of Putra Malaysia (UPM), in May 2003. The plant was identified by Mr.

Shamsul Khamis, a Resident Botanist at LHS, and a specimen (SK 1537/08) was deposited at the Herbarium of the Institute of Bioscience, UPM.

**Extraction and Isolation.** The dried plant material (16 kg) was extracted three times with distilled methanol, and the methanol extracts were combined and evaporated on a rotary evaporator. The crude methanolic extract (ca. 284 g) was then partitioned successively by solvent–solvent fractionation into four major fractions, *n*-hexane (60.31 g), ethyl acetate (69.43 g), *n*-butanol (72.61 g), and water (81.65 g). The butanolic extract (72.61 g) was loaded onto a Diaion HP-20 resin column and eluted with 100% water, water–methanol (3:1), water–methanol (2:1), water–methanol (1:1), water–methanol (1:2), and then 100% methanol, to yield six fractions altogether. The second and fourth fractions from the HP-20 column were combined and further fractionated into six subfractions using a ODS column and eluted with 100% water, water–methanol (3:1, 2:1, 1:1, 1:2), and 100% methanol. The subfractions were passed through a Sephadex LH-20 column again to obtain five subfractions with 100% water, water–methanol (2:1, 1:1, and 1:2), and 100% methanol sequentially. All of the subfractions thus obtained were purified by recycling HPLC (LC-908 on L-80 and GS-320 columns), with water–methanol (1:1) and methanol, respectively, as eluent, to obtain compounds **1** (2.7 mg, L-80, 5 mL/min, *t*<sub>R</sub> = 34 min), **2** (3.8 mg, GS-320, 5 mL/min, *t*<sub>R</sub> = 35 min), **3** (3.2 mg, L-80, 4.5 mL/min, *t*<sub>R</sub> = 44 min), **5** (4.1 mg, GS-320, 3 mL/min, *t*<sub>R</sub> = 54 min), **6** (3.7 mg, GS-320, 4 mL/min, *t*<sub>R</sub> = 32 min), **7** (1.1 mg, L-80, 4 mL/min, *t*<sub>R</sub> = 30 min), **8** (2.9 mg, L-80, 4 mL/min, *t*<sub>R</sub> = 37 min), **9** (4.3 mg, L-80, 4 mL/min, *t*<sub>R</sub> = 26 min), **10** (883 mg, L-80, 3 mL/min, *t*<sub>R</sub> = 38 min), **11** (1.15 g, L-80, 4 mL/min, *t*<sub>R</sub> = 72 min), **12** (125 mg, GS-320, *t*<sub>R</sub> = 44 min), **13** (862 mg, GS-320, 5 mL/min, *t*<sub>R</sub> = 38 min), borapetoside F (29 mg, L-80, 4 mL/min, *t*<sub>R</sub> = 70 min), and borapetoside G (3.1 mg, L-80, 4 mL/min, *t*<sub>R</sub> = 36 min), respectively.

The ethyl acetate fraction (69.43 g) was subjected to column chromatography (silica gel) and eluted by using gradients of *n*-hexane/acetone and *n*-hexane/ethyl acetate. Altogether, 15 fractions were collected. On further column chromatography of fraction 4 (26 mg), using 10% acetone–ethyl acetate as the mobile phase on normal-phase silica gel, compound **7** (2.0 mg) was obtained as a white solid.

**Hydrolysis of Compounds 1–9.** Compounds **1–9** (1 mg) were separately dissolved in 10% aqueous HCl (1 mL) at 90 °C for 4–5 h and cooled to room temperature. The reaction mixture was concentrated in vacuo, and each residue was subjected to TLC and GLC analysis in order to confirm the nature of the sugar moiety in the hydrolysate. TLC on silica gel developed with *n*-BuOH–Me<sub>2</sub>CO–H<sub>2</sub>O (4.5:1) indicated glucose at *R*<sub>f</sub> 0.33. The residue was trimethylsilylated with 0.1 mL of silylating agent (pyridine–trimethylchlorosilane, 5:1, by vol.) for 30 min at room temperature and subjected to GLC (1.5% silicon SE-30, 3 mm  $\times$  2 m, column temp 150 °C, N<sub>2</sub> 1.0 kg/cm<sup>2</sup>), which showed the presence of D-glucose with *t*<sub>R</sub> 10.4 min, [ $\alpha$ ]<sub>D</sub><sup>22</sup> +51.4 (*c* 0.02, H<sub>2</sub>O). Compound **1** showed the presence of D-xylose with *t*<sub>R</sub> 4.3 min, [ $\alpha$ ]<sub>D</sub><sup>22</sup> +19.1 (*c* 0.02, H<sub>2</sub>O), in addition to D-glucose.

**(2R,5R,6R,8S,9S,10S,12S)-15,16-Epoxy-2-hydroxy-6-O-( $\beta$ -D-glucopyranosyl)-(1 $\rightarrow$ 6)- $\alpha$ -D-xylopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester (**1**):** colorless, gummy solid; [ $\alpha$ ]<sub>D</sub><sup>27</sup> –35.0 (*c* 0.056, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 224 (3.68), 214 (4.03), 209 (4.27) nm; IR (KBr)  $\nu_{\max}$  3420, 1717, 1637 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Tables 1 and 3; EIMS *m/z* 373 (3), 133 (64), 119 (100), 95 (19), 94 (24), 91 (58), 81 (29), 73 (61), 60 (35), 55 (30); HRESIMS *m/z* 685.2701 [M + H]<sup>+</sup> (calcd for C<sub>32</sub>H<sub>45</sub>O<sub>16</sub>, 685.2708).

**(2R,5R,6R,8R,9S,10S,12S)-15,16-Epoxy-2-hydroxy-6-O-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester (**2**):** amorphous, colorless, gummy solid; [ $\alpha$ ]<sub>D</sub><sup>27</sup> –22.7 (*c* 0.17, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 223 (3.85), 217 (4.21), 201 (4.37) nm; IR (KBr)  $\nu_{\max}$  3409, 1709, 1670 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Tables 1 and 3; EIMS *m/z* 389 (13), 373 (37), 357 (20), 341 (27), 205 (31), 119 (70), 97 (29), 95 (66), 94 (64), 93 (23), 91 (68), 81 (100), 61 (53), 60 (60), 55 (62); HRESIMS *m/z* 553.2245 [M + H]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>37</sub>O<sub>12</sub>, 553.2280).

**(5R,6R,8S,9R,10R,12S)-15,16-Epoxy-2-oxo-6-O-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester (**3**):** amorphous, colorless, gummy solid; [ $\alpha$ ]<sub>D</sub><sup>27</sup> –33.3 (*c* 0.18, MeOH); UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 216 (4.55), 211 (4.55), 203 (4.55) nm; IR (KBr)  $\nu_{\max}$  3426, 1725, 1673 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Tables 1 and 3; EIMS *m/z* 550 (2), 388 (9), 371 (21), 356 (12), 203 (16), 167 (21), 133 (39), 119 (27), 95 (60), 94 (100), 93 (23), 91 (55), 81 (83),

73 (58), 61 (36), 60 (58), 55 (56); HRESIMS  $m/z$  551.2133 [M + H]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>35</sub>O<sub>12</sub>, 551.2123).

**Methyl (2R,7S,8S)-8-[(2S)-2-(3,4-dihydroxy-2,5-dimethoxytetrahydro-3-furanyl)-2-hydroxyethyl]-2,8-dimethyl-10-oxo-11-oxatricyclo[7.2.1.0<sup>2,7</sup>]dodec-3-ene-3-carboxylate (rumphiol E) (4):** amorphous, colorless solid;  $[\alpha]_D^{25} +27.5$  (c 0.04, CHCl<sub>3</sub>); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 242 (2.32) nm; IR (KBr)  $\nu_{max}$  3551, 3505, 3394, 1745, 1701 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CDCl<sub>3</sub>), see Table 4; EIMS  $m/z$  307 (94), 275 (47), 203 (14), 185 (21), 119 (16), 85 (100), 95 (8), 94 (95), 93 (29), 81 (12); HRESIMS  $m/z$  471.2312 [M + H]<sup>+</sup> (calcd for C<sub>23</sub>H<sub>35</sub>O<sub>10</sub>, 471.2225).

**(5R,6R,8S,9R,10S,12S)-15,16-Epoxy-2-oxo-6-O-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester (5):** amorphous, colorless, gummy solid;  $[\alpha]_D^{25} +260.8$  (c 0.046, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 225 (3.72), 216 (4.35), 205 (4.42) nm; IR (KBr)  $\nu_{max}$  3421, 1718, 1674 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Tables 2 and 3; HRESIMS  $m/z$  549.2171 [M + H]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>35</sub>O<sub>12</sub>, 549.2123).

**(2R,5R,6S,9S,10S,12S)-15,16-Epoxy-2-hydroxy-6-O-( $\beta$ -D-glucopyranosyl)-cleroda-3,7,13(16),14-tetraen-17,12-olid-18-oic acid methyl ester (6):** amorphous, colorless solid;  $[\alpha]_D^{25} +10.9$  (c 0.06, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 208 (3.82) nm; IR (KBr)  $\nu_{max}$  3430, 1725, 1664 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Table 4; EIMS  $m/z$  388 (9), 356 (6), 218 (38), 147 (45), 119 (43), 95 (64), 94 (60), 81 (71), 55 (100); FABMS  $m/z$  551 [M + H]<sup>+</sup>, 573 [M + Na]<sup>+</sup>, 389 [M - 162]<sup>+</sup>, 371 [M - 180]<sup>+</sup>; HRESIMS  $m/z$  573.1980 [M + Na]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>34</sub>O<sub>12</sub>Na, 573.1948).

**(5R,6S,9S,10S,12S)-15,16-Epoxy-2-oxo-6-O-( $\beta$ -D-glucopyranosyl)-cleroda-3,7,13(16),14-tetraen-17,12-olid-18-oic acid methyl ester (7):** amorphous, colorless solid;  $[\alpha]_D^{25} -14.2$  (c 0.01, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 210 (3.70), 201 (3.81) nm; IR (KBr)  $\nu_{max}$  3425, 1723, 1667 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Table 4; EIMS  $m/z$  386 (19), 354 (9), 309 (6), 216 (19), 202 (30), 145 (46), 135 (37), 95 (50), 94 (70), 81 (69), 73 (100); HRESIMS  $m/z$  549.1976 [M + H]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>33</sub>O<sub>12</sub>, 547.1967).

**(3R,4R,5R,6S,8R,9S,10S,12S)-15,16-Epoxy-3,4-epoxy-6-O-( $\beta$ -D-glucopyranosyl)-cleroda-3,13(16),14-trien-17,12-olid-18-oic acid methyl ester (8):** amorphous, gummy solid;  $[\alpha]_D^{25} +135.1$  (c 0.07, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 222 (4.46), 203 (4.65) nm; IR (KBr)  $\nu_{max}$  3447, 1714, 1635 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Tables 2 and 3; EIMS (70 eV)  $m/z$  (%) 373 (7), 341 (2), 323 (2), 205 (6), 194 (100), 119 (16), 95 (17), 94 (19), 93 (7), 91 (28), 81 (19), 73 (24), 61 (10), 60 (18), 59 (9), 55 (33); HRESIMS  $m/z$  553.2255 [M + H]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>37</sub>O<sub>12</sub>, 553.2280).

**(1R,4S,5R,8S,9R,10S,12S)-15,16-Epoxy-4-O-( $\beta$ -D-glucopyranosyl)-cleroda-2,13(16),14-triene-17(12),18(1)-diolide (9):** amorphous, colorless, gummy solid;  $[\alpha]_D^{25} -87.0$  (c 0.05, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\epsilon$ ) 224 (3.46), 213 (4.82), 207 (4.82) nm; IR (KBr)  $\nu_{max}$  3420, 1751, 1716 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR (CD<sub>3</sub>OD), see Tables 2 and 3; EIMS  $m/z$  314 (34), 281 (3), 252 (3), 220 (13), 206 (39), 134 (41), 121 (88), 119 (16), 108 (98), 107 (39), 97 (23), 95 (50), 94 (56), 93 (30), 91 (66), 81 (100), 73 (86), 60 (32), 55 (46); HRESIMS  $m/z$  521.2043 [M + H]<sup>+</sup> (calcd for C<sub>26</sub>H<sub>34</sub>O<sub>11</sub>, 521.2017).

**Cytotoxicity Assays.** Cytotoxicity of the compounds was evaluated in a 96-well flat-bottomed microplate by using a standard MTT (3-[4,5-dimethylthiazole-2-yl]-2,5-diphenyltetrazolium bromide) colorimetric assay.<sup>18</sup> For this purpose, PC-3 cells (human prostate cancer) and 3T3 cells (mouse fibroblasts) were cultured in Dulbecco's modified Eagle's medium (DMEM), supplemented with 5% fetal bovine serum (FBS), 100 IU/mL penicillin, and 100  $\mu$ g/mL streptomycin in a 25 cm<sup>3</sup>

flask, and kept in a 5% CO<sub>2</sub> incubator at 37 °C. Exponentially growing cells were harvested, counted with hemocytometer, and diluted with DMEM. Cell cultures at the concentration of 3 × 10<sup>4</sup> cells/mL were prepared and introduced (100  $\mu$ L/well) into 96-well plates. After overnight incubation, the medium was removed and 200  $\mu$ L of fresh medium was added with different concentrations of compounds (1–100  $\mu$ M). After 72 h, 50  $\mu$ L of MTT (2 mg/mL) was added to each well and incubated further for 4 h. Subsequently, 100  $\mu$ L of DMSO was added to each well. The extent of MTT reduction to formazan within cells was calculated by measuring the absorbance at 540 nm, using a microplate ELISA reader (Spectra Max Plus, Molecular Devices, CA). The cytotoxicity was recorded as the concentration causing 50% growth inhibition for both PC-3 and 3T3 cells. Doxorubicin (IC<sub>50</sub> 0.912 ± 0.12  $\mu$ M) and cycloheximide (IC<sub>50</sub> 0.26 ± 0.04  $\mu$ M) were used as positive control for the PC-3 and 3T3 cell lines, respectively.

**Acknowledgment.** We are grateful to the IDB (The Islamic Development Bank, Jeddah, Saudi Arabia) and COMSTECH, Islamabad, Pakistan, for financial support (M.I.C.) for the research project entitled, "Phytochemical, Pharmacological and Toxicological Studies on Medicinal Plants of Sudan and Pakistan".

**Supporting Information Available:** The <sup>1</sup>H, <sup>13</sup>C NMR, COSY, HMBC, HMQC, and NOESY/ROESY spectra of new compounds 1–9. These materials are available free of charge via the Internet at <http://pubs.acs.org>.

## References and Notes

- (1) Siddiqi, M. A.; Nasir, E.; Ali, S. I. *Flora of West Pakistan*; Ferozsons: Karachi, 1974; No. 74, p 5.
- (2) Noor, H.; Ashcroft, S. J. H. *J. Ethnopharmacol.* **1998**, *62*, 7–13.
- (3) Perry, L. M. *Medicinal Plants of East and South East Asia*; MIT Press: Cambridge, MA, 1980; p 268.
- (4) Herbal Medicine Research Group. *Compendium of Medicinal Plants Used in Malaysia*; Institute for Medical Research, Malaysia: Kuala Lumpur, 2002; Vol. 2, pp 392–393.
- (5) Singh, S. S.; Pandey, S. C.; Srivastava, S.; Gupta, V. S.; Patro, B.; Ghosh, A. C. *Indian J. Pharmacol.* **2003**, *35*, 83–91.
- (6) Chintalwar, G.; Jain, A.; Sipahimalani, A.; Banerji, A.; Sumariwalla, P.; Ramakrishnan, R.; Sainis, K. *Phytochemistry* **1999**, *52*, 1089–1093.
- (7) Shen, Y. C.; Cheng, Y. B.; Ahmed, A. F.; Lee, C. L.; Chen, S. Y.; Chien, C. T.; Kuo, Y. H.; Tzeng, G. L. *J. Nat. Prod.* **2005**, *68*, 1665–1668.
- (8) Maurya, R.; Wazir, V.; Kapil, R. S. *J. Indian Chem. Soc.* **1994**, *71*, 361–364.
- (9) Martin, T. S.; Ohtani, K.; Kasai, R.; Yamasaki, K. *Phytochemistry* **1995**, *40*, 1729–1736.
- (10) Atta-ur-Rahman; Ahmed, S.; Ali, S. S.; Shah, Z.; Choudhary, M. I. *Tetrahedron* **1994**, *50*, 12109–12112.
- (11) Fukuda, N.; Yonemitsu, M.; Kimura, T. *Liebigs Ann. Chem.* **1993**, 491–495.
- (12) Martin, T. S.; Ohtani, K.; Kasai, R.; Yamasaki, K. *Phytochemistry* **1996**, *42*, 153–158.
- (13) Akhila, A.; Rani, K.; Thakur, R. *Phytochemistry* **1991**, *30*, 2573–2576.
- (14) Manabe, S.; Nishino, C. *Tetrahedron* **1986**, *42*, 3461–3470.
- (15) Khan, M. A.; Gray, A. I.; Waterman, P. G. *Phytochemistry* **1989**, *28*, 273–275.
- (16) Pavia, D. L.; Lampan, G. M.; Kriz, G. S. *Introduction to Spectroscopy*; W. B. Saunders Co.: Philadelphia, 1979; pp 67–68.
- (17) Fukuda, N.; Yonemitsu, M.; Kimura, T.; Hachiyama, S.; Miyahara, K.; Kawasaki, T. *Chem. Pharm. Bull.* **1985**, *33*, 4438–4444.
- (18) Mosmann, T. *J. Immunol. Methods* **1983**, *65*, 55–63.

NP900551U